

## Communication

# Chiral Brønsted Acid-Catalyzed Tandem Aza-Ene Type Reaction/Cyclization Cascade for a One-Pot Entry to Enantioenriched Piperidines

Masahiro Terada, Kyoko Machioka, and Keiichi Sorimachi

*J. Am. Chem. Soc.*, **2007**, 129 (34), 10336-10337• DOI: 10.1021/ja0739584 • Publication Date (Web): 04 August 2007

Downloaded from http://pubs.acs.org on February 15, 2009



## More About This Article

Additional resources and features associated with this article are available within the HTML version:

- Supporting Information
- Links to the 54 articles that cite this article, as of the time of this article download
- Access to high resolution figures
- Links to articles and content related to this article
- Copyright permission to reproduce figures and/or text from this article

View the Full Text HTML





Published on Web 08/04/2007

#### Chiral Brønsted Acid-Catalyzed Tandem Aza-Ene Type Reaction/Cyclization Cascade for a One-Pot Entry to Enantioenriched Piperidines

Masahiro Terada,\* Kyoko Machioka, and Keiichi Sorimachi

Department of Chemistry, Graduate School of Science, Tohoku University, Sendai 980-8578, Japan

Received June 1, 2007; E-mail: mterada@mail.tains.tohoku.ac.jp

The development of efficient methods to access complex molecules with multiple stereogenic centers continues to be a substantial challenge in both academic research and industrial applications. One approach toward this challenge is the use of catalytic enantioselective cascade reactions,1 which have emerged as powerful tools to give a rapid increase in molecular complexity from simple and readily available starting materials, thus producing enantioenriched compounds in a single operation. It is obvious that such transformations require less solvents, adsorbents, and energy, hence minimizing waste management as compared to a series of individual stepwise reactions. In recent years, considerable efforts have been devoted to the development of enantioselective organic transformations using chiral Brønsted acids as green catalysts.<sup>2</sup> Cascade reactions catalyzed by chiral Brønsted acids are attractive because such catalytic processes would allow the production of enantioenriched compounds by ecologically and economically favorable methods.<sup>3</sup> In this communication, we describe a chiral monophosphoric acid<sup>4,5</sup>-catalyzed tandem aza-ene type reaction/ cyclization cascade that enables the rapid and highly enantio- and diastereoselective construction of piperidine derivatives with multiple stereogenic centers.

Recently, we successfully developed a highly efficient and enantioselective aza-ene type reaction<sup>6</sup> of N-acyl aldimines (1) with disubstituted enecarbamates (2:  $G \neq H$ ) catalyzed by chiral monophosphoric acid derivatives (3),<sup>4d</sup> in which the corresponding products (4) were obtained in ketimine form. Inspired by the formation of imines, we envisioned a sequential process using monosubstituted enecarbamates (5: G = H)<sup>4g,6e</sup> instead of the disubstituted versions (2). The acid-catalyzed reaction of initial aldimines (1) with 5 would generate aza-ene type products of *N*-acyl aldimines (6) as reactive intermediates and hence 6 would undergo further aza-ene type reactions leading to the subsequent generation of aldimines (7). If intramolecular cyclization of 7 could be enacted to terminate the tandem aza-ene type reaction sequence, our synthetic methodology would allow rapid access to piperidine derivatives (8) as key structural elements of numerous natural products (Scheme 1).<sup>7</sup>

*Scheme 1.* One-Pot Entry to Piperidine Derivatives via Tandem Aza-ene Type Reaction/Cyclization Cascade



10336 J. AM. CHEM. SOC. 2007, 129, 10336-10337

For the first step of the proposed cascade transformation, we examined the reaction of benzaldehyde-derived *N*-Boc aldimine (1a) with *N*-Cbz enecarbamate (5a) catalyzed by biphenol-derived phosphoric acid (9). To our delight, the cascade reaction proceeded smoothly to afford a diastereomeric mixture of the desired piperidine derivative (8a:  $R^1 = Ph$ ) quantitatively (eq 1). Although the diastereoselectivity was moderate, only two diastereomers were obtained from among the four possible diastereomers that result from three stereogenic centers.<sup>8</sup>



This preliminary result prompted us to develop enantio- and diastereoselective variants of our cascade transformation. Our studies were commenced with the screening of chiral phosphoric acid catalysts (3) bearing various type of aromatic substituents (Ar) at the 3,3'-position on the binaphthyl backbone. As shown in Table 1, all of the chiral catalysts (3) exhibited excellent performance both in terms of catalytic efficiency and enantioselectivity. The cascade reaction of **1a** with **5a** was completed within 30 min in the presence of (R)-3 (2 mol %) to afford the two diastereomers of the piperidine derivatives (*trans*- and *cis*-**8a**), as observed in the catalysis by **9** (entries 1–4 vs eq 1). Furthermore, excellent enantioselectivity was observed for the major *trans*-isomer, ir-

**Table 1.** Cascade Reaction of **1a** ( $R^1 = Ph$ ) with **5a** Catalyzed by (*R*)-**3** Leading to Piperidine Derivatives (**8a**) (eq 1)<sup>*a*</sup>

. ,	• ·				
entry	catalyst (3) (Ar)	yield (%)	trans:cis	ee (%) <sup>b,c</sup>	ee (%) <sup>b,d</sup>
1	3a (9-anthryl)	92	80:20	99	$12^e$
2	<b>3b</b> $(C_6H_5-)$	92	86:14	95	$19^e$
3	$3c (3,5-Ph_2-C_6H_3-)$	>99	86:14	96	31
4	$3d (4-Ph-C_6H_4-)$	>99	89:11	97	$8^e$
$5^{f}$	3d	>99	87:13	99	8
$6^g$	3d	>99	91:9	99	14
$7^h$	3d	97	92:8	99	14
$8^i$	3d	>99	95:5	>99	40

<sup>*a*</sup> Unless otherwise noted, all reactions were carried out with 0.10 mmol of **1a**, 0.21 mmol of **5a**, and 0.002 mmol of (*R*)-**3** (2 mol %) in 1.0 mL of CDCl<sub>3</sub> at room temperature for 30 min. <sup>*b*</sup> Enantiomeric excess was determined by chiral HPLC analysis. <sup>*c*</sup> % ee of *trans*-**8a** (absolute stereo-chemistry: 2R,4R,6S). <sup>*d*</sup> % ee of *cis*-**8a** (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R). <sup>*s*</sup> % ee of *cis*-*s* a (absolute stereo-chemistry: 2R,4R,6R).

Table 2. Scope of Substrates in Cascade Reaction of 1 with 5a Catalyzed by (R)-3d (eq 1)<sup>a</sup>

entry	1 (R1)	yield (%)	trans:cis	ee (%) <sup>b,c</sup>	ee (%) <sup>b,d</sup>
1	<b>1b</b> : <i>p</i> -Br-C <sub>6</sub> H <sub>4</sub> -	>99	94:6	99	23
2	1c: <i>p</i> -Me-C <sub>6</sub> H <sub>4</sub> -	>99	95:5	98	4
$3^e$	1d: 2-furyl	76	88:12	99	14
$4^e$	1e: Ph-CH=CH-	70	95:5	97	36
$5^e$	1f: $MeO_2C-$	84	88:12	98	$ND^{f}$
6 <sup>e</sup>	<b>1g</b> : <i>c</i> -C <sub>6</sub> H <sub>11</sub> −	68	94:6	97	48

<sup>a</sup> Unless otherwise noted, all reactions were carried out with 0.10 mmol of 1, 0.21 mmol of 5a, and 0.002 mmol of (R)-3d (2 mol %) in 1.0 mL of CH<sub>2</sub>Cl<sub>2</sub> at 0 °C for 5 h. <sup>b</sup> Enantiomeric excess was determined by chiral HPLC analysis. <sup>c</sup> % ee of trans-8. <sup>d</sup> % ee of cis-8. <sup>e</sup> Reaction was carried out using 0.005 mmol of (R)-3d (5 mol %). f Not determined % ee.

respective of the Ar substituent of the chiral catalysts (3) (entries 1-4). Interestingly, the simple phenyl-substituted catalyst (3b) also gave high enantioselectivity for the major trans-isomer (entry 2). In contrast, the diastereoselectivity was affected by the Ar substituents; among the catalysts examined, the para-biphenyl-substituted catalyst (3d) exhibited the highest diastereoselectivity (entry 4). Further optimization of reaction conditions by changing either solvents or reaction temperature was performed using 3d (entries 5-8). As a result, nearly enantiopure piperidine derivatives (8a) were obtained with high diastereoselectivity in CH2Cl2 at 0 °C (entry 8).

To investigate the scope of the present cascade transformations, the reaction of 5a with a series of aldimines (1) was examined using (R)-3d. Representative results are summarized in Table 2. It should be emphasized that, in most cases, one stereoisomer was formed exclusively from among the eight possible stereoisomers consisting of four pairs of enantiomers. Excellent enantioselectivities along with high diastereoslectivities were attained using aromatic aldimines (1b and 1c), regardless of their electronic nature (entries 1 and 2). Heteroaromatic and  $\alpha,\beta$ -unsaturated aldimines (1d and 1e) were also encouraging, giving the corresponding products (8d and 8e) in acceptable yields (entries 3 and 4). Moreover, the glyoxylate-derived aldimine (1f) could be transformed to the highly functionalized piperidine derivative (8f) in excellent enantioselectivity (entry 5). An aliphatic aldimine (1g) was also applicable to the present reaction, giving the product (8g) in high stereoselectivity (entry 6).

The high enantio- and diastereoselectivities observed can be attributed to the double asymmetric induction<sup>9</sup> arising from the matched combination between the optically active aldimine intermediates (6) and (R)-3. As shown in eq 1, catalysis of the cascade reaction by biphenol-derived phosphoric acid (9) resulted in 4,6trans selectivity, although the selectivity was moderate. The observed 4,6-trans diastereofacial selectivity induced by the chirality of 6 is identical to the stereochemical outcome in enantioselective catalysis, where (R)-3 preferentially directs attack of both the initial aldimines (1) and 6 onto the si face, giving a 4,6-trans relationship as the predominant relative stereochemistry. It is likely that the final step of the stereoselective cyclization proceeded under substrate control rather than enantioselective catalysis by (R)-3, because the reaction catalyzed by 9 also affords only two of the possible four diastereomers.

In summary, we have developed an efficient and highly diastereoand enantioselective tandem aza-ene type reaction/cyclization cascade, featuring a chiral monophosphoric acid catalyst, for a onepot entry to piperidine derivatives. With the control of three stereogenic centers, the cascade transformations can be widely applied using simple enecarbamates and a broad range of aldimines

to provide a rapid increase in molecular complexity. Further investigations of this cascade transformation are currently underway in our laboratory to elucidate the origin of the stereoselectivity and to construct more complex heterocyclic systems.

Acknowledgment. This work was supported by JSPS for a Grant-in-Aid for Scientific Research (B) (Grant No. 17350042) and The Tokuyama Science Foundation. We also acknowledge the JSPS Research Fellowship for Young Scientists (K.S.) from the Japan Society for the Promotion of Sciences.

Supporting Information Available: Representative experimental procedure, spectroscopic data for cascade reaction products (8), determination of relative stereochemistry of 8, and absolute stereochemistry of 8a. This material is available free of charge via the Internet at http://pubs.acs.org.

#### References

- For a recent review, see: Enders, D.; Grondal, C.; Hüttl, M. R. M. Angew. Chem., Int. Ed. 2007, 46, 1570–1581.
- (2) For reviews on chiral Brønsted acid catalysis, see: (a) Pihko, P. M. Angew. Chem., Int. Ed. 2004, 43, 2062-2064. (b) Taylor, M. S.; Jacobsen, E. N. Angew. Chem., Int. Ed. 2006, 45, 1520-1543. (c) Akiyama, T.; Itoh, J.; Fuchibe, K. Adv. Synth. Catal. 2006, 348, 999-1010. (d) Connon, S. J. Angew. Chem., Int. Ed. 2006, 45, 3909-3912.
- (3) For selected examples of organocatalytic tandem reactions, see: (a) Yamamoto, Y.; Momiyama, N.; Yamamoto, H. J. Am. Chem. Soc. 2004, J.Zó, 5962, -5963. (b) Marigo, M.; Schulte, T.; Franzen, J.; Jørgensen, K.
   A. J. Am. Chem. Soc. 2005, 127, 15710–15711. (c) Casas, J.; Engqvist, A. J. Am. Chem. Soc. 2005, 127, 15110
   H. G. Casas, S., Elgylst,
   M.; Ibrahem, I.; Kaynak, B.; Córdova, A. Angew. Chem., Int. Ed. 2005, 44, 1343–1345. (d) Yang, J. W.; Hechavarria, Fonseca, M. T.; List, B. J. Am. Chem. Soc. 2005, 127, 15036–15037. (e) Huang, Y.; Walji, A. M.; Larsen, C. H.; MacMillan, D. W. C. J. Am. Chem. Soc. 2005, 127, 15051–15053. (f) Boxer, M. B.; Yamamoto, H. J. Am. Chem. Soc. 2005, 128, 48, 400 (c) Ware, V. Lie, V. Dera, L. L. M. Chem. Soc. 2006, 128, 48, 400 (c) Ware, V. Lie, V. Dera, L. L. M. Chem. Soc. 2006, 128, 120 (c) Ware, M. B.; Yamamoto, H. J. Am. Chem. Soc. 2005, 128, 120 (c) Ware, 2006, 128, 1905 (f) Wang, Y. Liu, X.; Deng, L. J. Am. Chem. Soc. 2006, 126, 3928–3930. (h) Enders, D.; Hüttl, M. R. M.; Grondal, C.; Raabe, G. Nature 2006, 441, 861-863. (i) Brandau, S.; Maerten, E.; Jørgensen, K. A. J. Am. Chem. Soc. **2006**, 128, 14986–14991. (j) Zu, L; Wang, J;; Li, H.; Xie, H.; Jiang, W.; Wang, W. J. Am. Chem. Soc. **2007**, 129, 1036– 1037
- (4) (a) Uraguchi, D.; Terada, M. J. Am. Chem. Soc. 2004, 126, 5356-5357. (a) Olaguchi, D., Ferada, M. J. Am. Chem. Soc. 2004, 120, 5350
   (b) Uraguchi, D.; Sorimachi, K.; Terada, M. J. Am. Chem. Soc. 2004, 126, 11804–11805. (c) Uraguchi, D.; Sorimachi, K.; Terada, M. J. Am. Chem. Soc. 2005, 127, 9360–9361. (d) Terada, M.; Machioka, K.; Sorimachi, K. Angew. Chem., Int. Ed. 2006, 45, 2254–2257. (e) Terada, M.; Sorimachi, K.; Uraguchi, D. Synlett 2006, 133–136. (f) Gridnev, I. D.; Kouchi, M.; Sorimachi, K.; Terada, M. Tetrahedron Lett. 2007, 48, 497-500. (g) Terada, M.; Sorimachi, K. J. Am. Chem. Soc. 2007, 129, 292-293. (h) Terada, M.; Yokoyama, S.; Sorimachi, K.; Uraguchi, D.
- 292-295. (f) Terada, M.; Fokoyama, S.; Sorimacin, K.; Oraguchi, D. Adv. Synth. Catal. 2007, in press.
  (5) (a) Akiyama, T.; Itoh, J.; Yokota, K.; Fuchibe, K. Angew. Chem., Int. Ed. 2004, 43, 1566-1568. (b) Rowland, G. B.; Zhang, H.; Rowland, E. B.; Chennamadhavuni, S.; Wang, Y.; Antilla, J. C. J. Am. Chem. Soc. 2005, 127, 15696-15697. (c) Storer, R. I.; Carrera, D. E.; Ni, Y.; MacMillan, D. W. C. J. Am. Chem. Soc. 2006, 128, 84-86. (d) Seayad, L.; Sozud, A. M.; List, P. J. Am. Chem. Soc. 2006, 128, 1086-1087 (c) J.; Seayad, A. M.; List, B. J. Am. Chem. Soc. 2006, 128, 1086-1087. (e) K. Buejing, M.; Kuchak, M.; List, D. S. Mi, Chem. Jour. 2006, 120, 1000 (100); ( T.; Morita, H.; Fuchibe, K. J. Am. Chem. Soc. 2006, 128, 13070-13071. (i) Chen, X.-H.; Xu, X.-Y.; Liu, H.; Cun, L.-F.; Gong, L.-Z. J. Am. Chem.
   (j) Chen, X.-H.; Xu, X.-Y.; Liu, H.; Cun, L.-F.; Gong, L.-Z. J. Am. Chem.
   Soc. 2006, 128, 14802–14803. (k) Kang, Q.; Zhao, Z.-A.; You, S.-L. J.
   Am. Chem. Soc. 2007, 129, 1484–1485. (l) Li, G.; Liang, Y.; Antilla, J.
   C. J. Am. Chem. Soc. 2007, 129, 5830–5831. (m) Rueping, M.;
   Ieawsuwan, W.; Antonchick, A. P.; Nachtsheim, B. J. Angew. Chem., Int.
   C. J. Constant of the state of the stat Ed. 2007, 46, 2097-2100. (n) Yamanaka, M.; Itoh, J.; Fuchibe, K.;
- Ed. 2007, 46, 2097–2100. (n) Yamanaka, M.; Itoh, J.; Fuchibe, K.;
  Akiyama, T. J. Am. Chem. Soc. 2007, 129, 6756–6764.
  (a) Matsubara, R.; Nakamura, Y.; Kobayashi, S. Angew. Chem., Int. Ed.
  2004, 43, 1679–1681. (b) Matsubara, R.; Nakamura, Y.; Kobayashi, S.
  Angew. Chem., Int. Ed. 2004, 43, 3258–3260. (c) Matsubara, R.; Vital,
  P.; Nakamura, Y.; Kiyohara, H.; Kobayashi, S. Tetrahedron 2004, 60,
  9769–9784. (d) Fossey, J. S.; Matsubara, R.; Vital, P.; Kobayashi, S.
  Ore Biomod. Chem. 2005, 3 2010–2012. (c) Matsubara P.; Kobayashi, S. (6)Org. Biomol. Chem. 2005, 3, 2910-2913. (e) Matsubara, R.; Kawai, N.; Kobayashi, S. Angew. Chem., Int. Ed. 2006, 45, 3814-3816. (f) Kiyohara, H.; Matsubara, R.; Kobayashi, S. Org. Lett. 2006, 8, 5333–5335.
   (7) Bailey, P. D.; Millwood, P. A.; Smith, P. D. Chem. Commun. 1998, 633-
- The relative stereochemistries were determined by NOE experiments.
- Masamune, S.; Choy. W.; Petersen, J. S.; Sita, L. R. Angew. Chem., Int. Ed. Engl. 1985, 24, 1–31.
  - JA0739584